

Determination of allantoin in blood by high-performance liquid chromatography with pre-column derivatization

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ABSTRACT

A high-performance liquid chromatography method with pre-column derivatization for separation and quantification of allantoin in blood samples is described. Plasma after deproteinisation with trichloroacetic acid was used for the derivatization procedure. The procedure was based on allantoin conversion to glyoxylic acid which forms a hydrazone with 2,4-dinitrophenylhydrazine. Allantoin derivatives (*syn- and anti-isomers*) were separated on a reversed phase column (Nova-Pak C₁₈, 4 µm) by gradient elution, and then monitored at 360 nm. All components were completely resolved in about 46 min. The average recovery of allantoin added to plasma samples was $101.3 \pm 8.7\%$ ($n = 52$). With UV detector the smallest allantoin concentration that gave reproducible integrations was $0.93 \mu\text{mol/l}$. The within-assay coefficient of variation CV for derivatization and injection was $2.7 \pm 1.2\%$, while CV for repeated injections was $0.68 \pm 0.33\%$. This HPLC method can also be used for determination of allantoin in urine.

KEY WORDS: allantoin, blood, determination, HPLC

INTRODUCTION

In ruminants purines are metabolized in a series of reactions to form allantoin, uric acid, hypoxanthine and xanthine. Allantoin excreted in the urine is the main end product of purine metabolism (Balcells et al., 1991; Watts, 1980) and

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originated from three possible sources: purine bases of rumen microorganisms, endogenous purines from tissue turnover, and feed purines. In sheep and other ruminants allantoin appears to originate predominantly from nucleic acids synthesized by rumen microorganisms (Antoniewicz et al., 1980). In ruminants there may be a close relationship between the production of microbial protein in the rumen and the excretion of purine derivatives in urine and blood. Thus, measurements of the allantoin level in blood and urine provides an indice of the amount of microbial biomass supplied to ruminants (Ryś et al., 1975; Antoniewicz et al., 1980, 1981; Lindberg et al., 1989; Chen et al., 1992). Several authors have suggested that appropriate indicator of microbial protein synthesis may be the excretion of total purine derivatives (Chen et al., 1990; Puchała et al., 1991).

One major advantage is that this approach does not require the use of an invasive method to estimate microbial protein supply. Therefore it is necessary to resolve problem of determination of allantoin level in physiological fluids (Balcells et al., 1992; Terzuoli et al., 1994, 1995). However the method based on measurements of allantoin content in urine requires a total collection of urine (e.g. 1-3 l/day for sheep) and good separation of urine from faeces is essential (Lindberg et al., 1989; Gonda et al., 1994). Considering the above facts, it is essential to provide an accurate and selective method for the determination of allantoin in blood offering satisfactory estimation of the size of bacterial protein synthesis.

The aim of our work was to examine the suitability and accuracy of a pre-column derivatization (Young et al., 1942; Chen et al., 1993) and separation by HPLC for determination of allantoin in blood.

MATERIAL AND METHODS

Reagents

HPLC-grade methanol, acetonitrile were purchased from Merck (Darmstadt, Germany), allantoin and 2, 4-dinitrophenylhydrazine (DNPH) from Sigma (St. Louis, MO, USA). All other chemicals were of analytical reagent grade and purchased from POCH (Gliwice, Poland). Water was distilled and then deionized prior to use. HPLC-grade water was prepared using a Milli-Q system (Millipore, Toronto, Canada). Mobile phases (solvents A and B) were filtered through a 0.2 μm membrane filter (Millipore). A degassing of the solvents was made by 15 min ultrasonication prior to use.

HPLC configuration

A Waters 625 LC HPLC system (including a controller and pumps) was employed. The apparatus consisted of a turnable absorbance detector Waters Model 486, Waters 712 WISP autosampler, and computer data handling system (all equipment from Waters, Millipore, MA, USA).

Analytical method development, collection and data integration were performed by using Millennium 2001 software on a PC-386 computer. Chromatography was carried out using a Nova-Pak C₁₈ (4 μm), 15 mm x 183.9 mm I.D. (Waters, Millipore). A guard column (Waters, Millipore) of 10 x 6 mm I.D. packed with reversed phase C₁₈ (30-40 μm) pellicular packing material was used.

Analytical solvents and gradient composition

The derivatizing solution was prepared by dissolving 100 mg of DNPH in 100 ml of 2 M HCl and further the obtained solution was filtered. Since the commercial DNPH contains 30% water, the actual concentration was 3.5 M. Thymol blue (pH indicator) was prepared at a concentration of 0.04% w/v.

The binary gradient program was used (Waters curvilinear program). The solvent A was 9% methanol in 0.01 acetic acid (v/v) (adjusted with ammonia solution to pH 6.1) and solvent B was acetonitrile. The gradient used was (% B): 0% B at 0 min, 5% at 21 min (concavely increased from 17 min, line No. 7), 11% at 39 min (concavely increased, line 7), 0% at 41 min (linearly decreased from 39 min, line No. 6) and 0% at 47 min (line 11). Flow rate was 1 ml/min (the system pressure 2150 ± 40 psi). Injection volume was 10 μl. Detector was set at UV with an attenuation of 0.050 a.u.f.s., time was 47 min. All separation was performed at room temperature. To provide calibration for blood samples, a set of allantoin standards (6.7-401.6 μmol/l) was used.

Blood samples preparation

Blood samples from jugular vein of sheep were collected into tubes containing heparin and centrifuged at 1500 g for 20 min. The plasma was stored at 20°C. On the day of analysis, 0.5 ml of plasma was deproteinized with 0.5 ml of 10% (w/v) trichloroacetic acid and centrifuged at 10000-12000 rpm for 15 min. The supernatant was used for the derivatization procedure.

Derivatization procedure

A 500 μl sample of deproteinized plasma or allantoin standards and 50 μl of the pH indicator were pipetted into a centrifuge tube. To plasma samples 280 μl

of 0.6 M NaOH was added while to allantoin standards 100 μl of 0.6 M NaOH only. The mixture could be stored at about -10°C if necessary. After heating at 85°C for 60 min, 200 μl of the DNPH solution was added. Next, a sample was filtered through 0.2 μm filter (Cole Parmer) into the autosampler vial.

RESULTS AND DISCUSSION

The major analytical problem in the present work was obtaining a suitable separation of allantoin-deriving products from the interfering compounds. The derivatization procedure, however is not specific due to interference by 2, 4-dinitrophenylhydrazone of other keto-acids in plasma samples. Moreover, plasma contain several components with a similar high polarity and UV absorption. To omit these problems the method was improved.

The effect of different mobile phases composition and applied gradient in the separation of allantoin derivatives (i.e. *syn* and *anti* isomers: allantoin-A and allantoin-B peaks) is shown in Figure 1. As can be seen from the chromatograms in Figure 1a and 1c, plasma samples contain peaks (I and II) of unidentified DMPH complexes that had areas comparable to those of allantoin derivative peaks (A, B). The unidentified substances are presumably hydrazones of other keto-acids (Chen et al., 1993). As expected, the improved HPLC method enabled suitable separation of analytical peaks of allantoin A and B. Indeed, the peaks A and B were distinct from interfering substances in plasma, so, their presence do not affect the determination of allantoin.

Although no reduction in a area of peak A or B was observed, the increase in a peak area of a unidentified compound (I) (the retention time of 15.2 ± 0.4 min; Figure 1a) was found (Table 1) when the processed plasma samples were stored for 3 days at room temperature.

The peaks A and B had a retention time of 18 ± 2 and 43 ± 2 min (mean \pm SD of 90 samples), respectively. The ratio of allantoin-B/A peak area was constant at 3.406 ± 0.196 (calculation based on 60 samples), irrespective of the concentration of allantoin and type of sample. Obviously, peaks A and B were suitable for quantitative determination of the unknown, although the larger peak A was preferred since more precise and accurate results were achieved. Thus, all determinations of allantoin concentration were based on allantoin-A peak.

Accuracy of the method was assessed by examining the recovery of known quantities (27.0-229.7 $\mu\text{mol/l}$) of allantoin added to plasma samples and was on the average $101.2 \pm 8.7\%$ (Table 2) .

The relationship between the concentration (y) and allantoin-A peak area (S_N) was linear over a wide range of allantoin content (6.7-401.6 $\mu\text{mol/l}$)

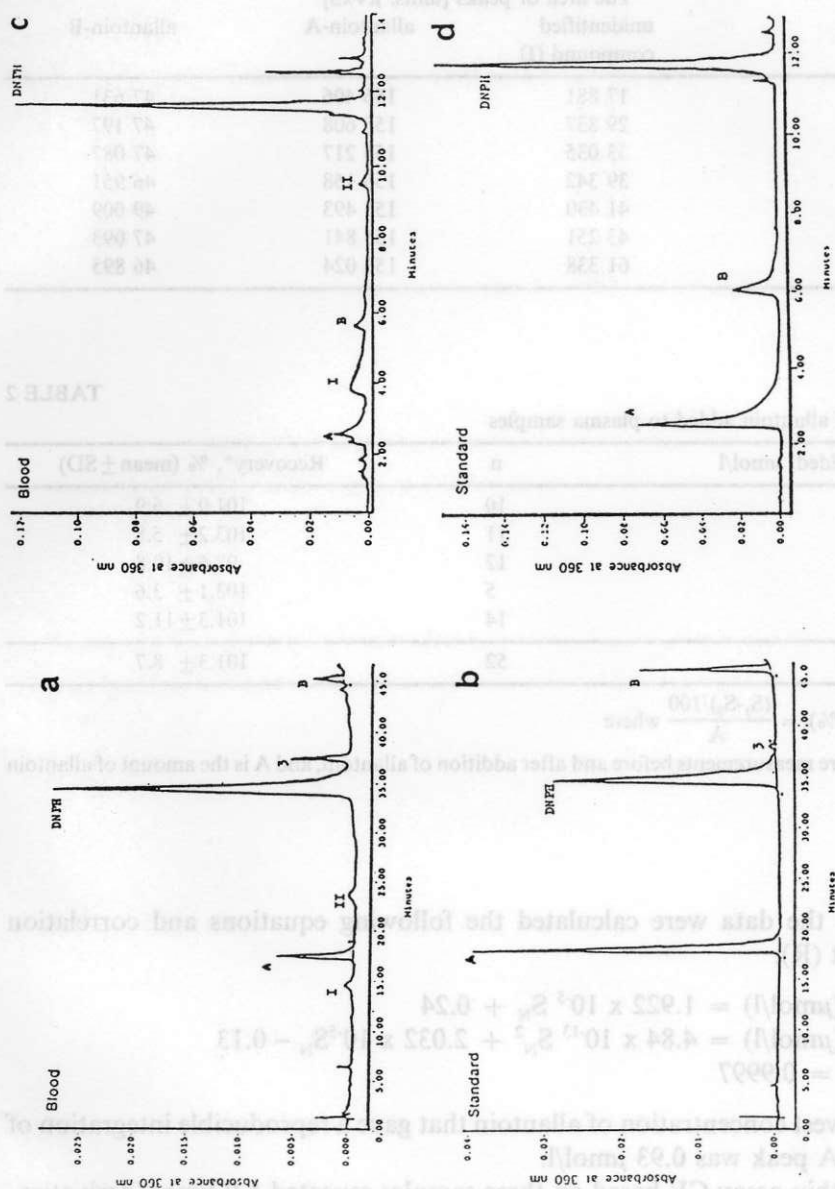


Figure 1. Comparison of improved method with Chen's method (Chen et al., 1993). Chromatograms for plasma sample (a) and allantoin standards (b) by improved method. Chromatograms for plasma sample (c) and allantoin standards (d) by the Chen's method. Peaks A and B are allantoin A and B isomers of 2,4-dinitrophenylhydrazone of glyoxylic acid. Peaks I, II and 3 are two unidentified compounds and thymol blue (at 37 ± 1 min), respectively

TABLE 1
Effect of length of plasma storage at room temperature on levels of an unidentified compound (I) and allantoin derivatives

Time	The area of peaks [units: $\mu\text{V}\times\text{S}$]		
	unidentified compound (I)	allantoin-A	allantoin-B
10 min	17 881	156 406	47 631
50 min	29 837	155 608	47 197
100 min	33 035	155 217	47 087
150 min	39 342	155 158	46 951
200 min	41 430	155 493	49 009
250 min	43 251	155 841	47 093
3 days	61 338	158 024	46 895

TABLE 2
Recovery of allantoin added to plasma samples

Allantoin added, $\mu\text{mol/l}$	n	Recovery*, % (mean \pm SD)
27.0	10	101.9 \pm 6.9
53.0	11	103.2 \pm 5.8
102.1	12	98.6 \pm 10.8
125.3	5	103.1 \pm 3.6
229.7	14	101.3 \pm 11.2
Pooled data	52	101.3 \pm 8.7

* recovery (%) = $\frac{(S_1 - S_0)/100}{A}$ where

S_0 and S_1 are measurements before and after addition of allantoin, and A is the amount of allantoin added.

and from the data were calculated the following equations and correlation coefficient (R):

$$y (\mu\text{mol/l}) = 1.922 \times 10^{-5} S_N + 0.24$$

$$y (\mu\text{mol/l}) = 4.84 \times 10^{-13} S_N^2 + 2.032 \times 10^{-5} S_N - 0.13$$

$$R = 0.9997$$

The lowest concentration of allantoin that gave a reproducible integration of allantoin-A peak was 0.93 $\mu\text{mol/l}$.

The within-assay CV based on three samples repeated 5-9 times (derivatization and injection) was $2.7 \pm 1.2\%$ ($n=20$). The CV for repeated injections was $0.68 \pm 0.33\%$ based on three samples each with six injections.

CONCLUSION

The present method is highly selective for allantoin, and offers the necessary sensitivity to permit the determination of allantoin in blood samples. Obviously, this HPLC method can also be used for determination of allantoin in urine. The chromatographic separation uses a widely available reversed-phase C₁₈ column. Due to the complete separation of allantoin derivatives, the proposed method has advantages over Chen's method (Chen et al., 1993). The allantoin derivatives are stable, however, increase of concentrations of some unidentified compounds was observed when the processed blood samples were stored for 3 days at room temperature. Thus, plasma samples should be subjected to chromatographic separation no later than after 3 days storage at room temperature following the derivatization procedure.

The run time of 46 min appears long, but information provided by plasma samples are worth the effort. Indeed, application of this HPLC method in the study of purine metabolism should provide further evidence of the influence of nutritional manipulation on microbial supply protein to the small intestine.

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STRESZCZENIE

Oznaczanie allantoiny we krwi metodą HPLC po wstępnym przeprowadzeniu jej w pochodną

Opisano metodę oznaczania allantoiny we krwi przy użyciu zestawu HPLC, po przeprowadzeniu jej w związek pochodny, pozwalającą na rozdzielenie od substancji interferujących. Allantoinę w odbiałczonym kwasem trójchlorooctowym osoczu krwi przeprowadzono w jej pochodną – kwas glikosalowy, tworzący z kolei hydrazon z 2, 4-dinitrofenylohydrazyną. Tę pochodną allantoiny (*syn* i *anty* izomery) rozdzielono na kolumnie z odwróconą fazą (Nova-Pak C₁₈, 4 μm) poprzez elucję gradientową i monitorowanie przy 360 nm. Czas rozdziału wszystkich składników wynosił około 46 min. Odzysk allantoiny dodanej do osocza wynosił $101.3 \pm 8.7\%$ (n = 52). Najmniejsze stężenie allantoiny pozwalające na powtarzalną integrację wynosiło 0,93 μmola/l. Współczynnik zmienności wyników analizy (CV) w obrębie przeprowadzania w pochodne i iniekcji prób wynosił $2,7 \pm 1,2\%$, podczas gdy CV dla powtórzeń iniekcji wynosił $0,68 \pm 0,33\%$. Metoda ta może służyć także do oznaczania allantoiny w moczu.